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- Manuscripts should be printed with a font size of 11.5 or 12 (also References, Tables and Figure legends) and preferably in 25 lines per page with 2.5 cm-wide margins on the right and left sides of the paper. Please do not justify the right margin.
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- 4. The paper should not form part of a numbered series.
- 5. For key words, general terms such as 'Membrane', 'Structure', 'Enzyme', 'Inactivation', etc. should not be used. Instead, e.g. 'Membrane structure', 'Enzyme inactivation' are appropriate.
- 6. Line drawing should be in black ink on white paper. Half-tone photograph should be submitted as sharp and contrasting glossy prints. Legends for these must be typed on a separate sheet. These should be of a quality suitable for direct repro

duction and approximately twice the size that they will appear. The preferable width of printed size for column is within 75mm. The maximum printed size for page is 165mm \times 220 mm. Figures and tables should also be compiled on separate sheets, each of which author (s)' names and the number were noted with pencil.

- 7. For nomenclature, abbreviations and units authors should follow internationally accepted general rules. See Tables 1, 2, and 3.
- 8. References in the text should be cited as follows: for two authors, Hugo and Franklin (1968) ; for three or more authors, Reid et al. (1995). References to papers by the same author (s) in the same year should be distinguished by the letters, a, b, c, etc. (for example, 1995a). References at the end of the paper must be given in alphabetical order, except for papers with three or more authors, which should be listed in chronological order after any other papers by the first author. See examples below for references to journals and books.
 - Hugo, W. B., and Franklin, I. (1968) Cellular lipid and the antistaphylococcal activity of phenols. *J. Gen. Microbiol.*, **52**, 365-373.
 - Kourai, H., Manabe, Y., Matsutani, E., Hasegawa, Y., and Nakagawa, K. (1995) Antimicrobial activities of alkylallyldimethylammonium iodides and alkylallyldiethylammonium iodides. *J. Antibact. Antifung. Agents*, **23**, 271-280. (Before 1996)
 - Ueda, S., Mineno, J., and Kuwabara, Y. (1999) Evaluation of the PCR method for the detection of verotoxin-producing *Escherichia coli* in foods and other meterials (in Japanese). *Bokin Bobai*, **27**, 441-446. (After 1997)
 - Reid, G., Khoury, A., and Nickel, J. C. (1991) The process of microbial biofilm formation in medical devices. In *Biodeterioration and Biodegradation 8* (Rossmoore, H. W., ed.), pp. 187-195, Elsevier Science Publishers, New York.
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1) The papers intended for the human body shall be the studies conducted with respect for "the Declaration of Helsinki" (1964, Revised 2008), "Ethical Guidelines

for Epidemiological Studies" by the Ministry of Education/ the Ministry of Health, Labor and Welfare; "Ethical Guidelines for Clinical Studies" by the Ministry of Health, Labor and Welfare; "Ethical Guidelines for Studies on the Human Genome and Gene Analyses" by the Ministry of Education / the Ministry of Health, Labor and Welfare / the Ministry of Economy, Trade and Industry and other similar ethical guidelines. In either case, please add a note, in a submitted paper, that the study has been conducted with the approval of the Ethical Review Board of an affiliated organization.

2) The papers dealing with animals shall be the studies that have been conducted with respect for the guideline for animal experiments specified by the Ethical Review Board of the affiliated organization based on the intention of "Act on Welfare and Management of Animals"; "Basic Guidlines for Conducting Animal Experiments in the Organization Under the Jurisdiction of the Ministry of Health, Labour and Welfare"; and "Standard for Breeding/Safekeeping of Animals for Experiments and the Reduction of Pain" by the Ministry of the Environment. Please add a note, in a submitted paper, that the study has been conducted with the approval of the Ethical Review Board of an affiliated organization.

Quantity	Symbol	Unit
Time	t	yr (not year), mo (not month), wk (not week), d (not day), h (not hour), min, s (not second),
Longth	1	ms, μ s, ns
Length	l A	m, mm, μ m (not μ), nm m ² cm ² mm ² μ m ² nm ²
Area Volume	A V	m^2 , cm ² , mm ² , μm^2 , nm ²
volume	V	m ³ , dm ³ , cm ³ , mm ³ , μ m ³ , nm ³
Mass	m	kl, <i>l</i> , ml, μ l, nl kg, mg, μ g (not γ), ng
Concentration	m C	M, mM, μ M, nM
Amount of substance		
Molecular mass	n	mol, mmol, μ mol, nmol
	m Mr	Da (dalton)
Relative molecular mass	Mr	dimensionless
Molar mass	M T	g·mol ^{−1} K, ℃
Temperature		
Heat	<i>q</i> , Q	kJ, J (not cal)
Electricity and magnetism Electric current	1	A mA // A
Potential difference	V	A, mA, μ A
	-	V, mV, μ V
Capacitance	С	F (farad)
Magnetic field strength	H	G (gauss)
Resistance	R	Ω
Conductivity	K	$S(\Omega^{-1})$
Force	F	N (kg·m·s ⁻²) P_{2} (N m^{-2})
Pressure	Р	Pa $(N \cdot m^{-2})$, atm, bar, mbar, torr, kg/cm ² , mm H
Sedimentation coefficient	S	$S (=10^{13}s)$
Density	ρ	g·cm ^{−3}
Relative density	d	
Viscosity	η	P (=0.1 Pa·s)
Frequency	V , f	Hz
Wavelength	λ	nm (not m μ)
Absorbance	A	dimensionless
Transmittance	Т	dimensionless $M^{-1} = m^{-1}$
Molar absorption coefficient	3	$M^{-1} \cdot cm^{-1}$
Ionic strength		$M (mol \cdot l^{-1}), mM$
Energy	E	J (not cal)
Gibbs free energy	G	J (not cal)
Equilibrium constant	K	dimensionless
Michaelis constant	Km	M, mM
Inhibition constant	Ki	M, mM
Rate constant	k	s^{-1} , $M \cdot s^{-1}$
Rate of reaction	V	$mol \cdot s^{-1}$, $mmol \cdot s^{-1}$
Other units		
Curie		Ci
Roentgen		R

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(cont'd)

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Acceleration of gravity	g	
Prefixes for units \therefore E, exa (10 ¹⁸);P,	peta (10^{15}) : T, tera (10^{12}) ; G, giga (10^{9}) ; M,
mega (10^6) ; k, kilo (10^3) ; h,	hecto	(10^2) ; da, deca (10^1) ; d, deci (10^{-1}) ; c, centi
(10^{-2}) ; m, milli (10^{-3}) ; μ ,	micro	(10^{-6}) ; n, nano (10^{-9}) ; p, pico (10^{-12}) ; f,
femto (10^{-15}) ; a, atto (10^{-18})		

mical compounds.	

TABLE 2. Abbreviations for words other than units, quantities, or chen

biological oxygen demand

BOD

БОЛ	biological oxygen demand
b.p.	boiling point
calc.	calculated
cf.	compare
COD	chemical oxygen demand
conc.	concentrated
concn	concentration
cpm	counts per minute
dil.	dilute
dpm	disintegration per minute
e.g.	for example
et al.	et alia (and others)
Expt. (pl. Expts.)	Experiment(s)
Fig. (pl. Figs.)	Figure(s)
i.e.	that is
max.	maximum
min.	minimum
m-, p-, o-	meta-, para-, ortho-
m.p.	melting point
n-	normal
no.	number
%	percent
p. (pl. pp.)	page(s)
ppb	parts per billion
ppm	parts per million
ppt.	precipitate
ref. (pl. refs.)	reference(s)
R _f	distance traveled by zone, divided by distance traveled by solvent
	front
rpm	revolutions per minute
S.D.	standard deviation
S.E.	standard error
sec-	secondary
spec. act.	specific activity
temp.	temperature
tert- (t-)	tertiary
UV	ultraviolet
vol. (pl. vols.)	volume(s)
VS.	versus
v/v	volume : volume
vvm	volume per volume per minute
wt.	weight
w/v	weight : volume
w/w	weight : weight

ADP	adenosine 5'-diphosphate
AMP	adenosine 5'-monophosphate
ATP	adenosine 5'-triphosphate
ATPase	adenosine triphosphatase
CoA	coenzyme A
DNA	deoxyribonucleic acid
cDNA	complementary DNA
DNase	deoxyribonuclease
EDTA	ethylenediaminetetraacetic acid
FAD	flavin adenine dinucleotide
FMN	flavin mononucleotide
NAD, NAD $^+$	nicotinamide adenine dinucleotide
NADH	reduced NAD
NADP, NADP ⁺	nicotinamide adenine dinucleotide phosphate
NADPH	reduced NADP
RNA	ribonucleic acid
mRNA	messenger RNA
rRNA	ribosomal RNA
tRNA	transfer RNA
RNase	ribonuclease
SDS	sodium dodecylsulfate
Tris	Tris (hydroxymethyl) aminomethane

TABLE 3. Abbreviations for chemical compounds (these abbreviations may be used without definition).

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